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# A Ca<sup>2+</sup>-STIMULATED INCORPORATION OF PHOSPHATE INTO ATP IN CHLOROPLASTS; THE PROBLEM OF ALLOTOPY

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#### Summary

Under continuous illumination, a  $Ca^{2+}$ -induced incorporation of phosphate into ATP is found in chloroplasts, with properties similar to the  $Mg^{2+}$ -catalyzed ATP— $P_i$  exchange. Thus, the chloroplast coupling factor has no allotopic properties towards divalent cations.

Until recently, the behaviour of chloroplast coupling factor 1 (CF<sub>1</sub>) towards divalent cations was classified as 'allotopic' [1]: the activities of the membrane-bound form, like photophosphorylation and light-induced ATPase, were  $Mg^{2+}$ -dependent [2,3], whereas the ATPase of purified CF<sub>1</sub> was dependent on the presence of  $Ca^{2+}[4]$ . Two years ago Nelson et al. [5] showed that isolated CF<sub>1</sub>, under appropriate conditions, could also exhibit  $Mg^{2+}$ -catalyzed ATPase.

Considerations of symmetry would require the existence of membrane-bound Ca<sup>2+</sup>-dependent activities as well. It occurred to us that the reason why these had not been observed might be the unique energetic feedback features of the system: chloroplast ATPase characteristically requires a high-energy conformation, and during Mg<sup>2+</sup>-ATPase, this conformation under suitable conditions can be maintained by the ATPase reaction itself [6–8]. In other words, Mg<sup>2+</sup> fulfills at least a dual function here: as a cofactor in the turnover of the ATPase enzyme, and in coupling it to the formation of the high-energy state required for maintaining that turnover. It might very well be that Ca<sup>2+</sup> could substitute for Mg<sup>2+</sup> in the first function, like it does with

TABLE I COMPARISON OF  $Mg^{2+}$  AND  $Ca^{2+}$  -STIMULATED PHOSPHATE INCORPORATION IN CHLOROPLASTS

Spinach chloroplasts were prepared as in ref. 7, but in a medium containing 100 mM KCl, 0.5 mM Na-EDTA and 5 mM Na-tricine (pH 8.2). The reaction mixture contained, in addition, 10  $\mu$ M pyocyanin, 3 mM  $P_i$  containing about 10<sup>6</sup> cpm of <sup>32</sup> $P_i$ , and 10 mM DTE and 5 mM MgCl<sub>2</sub> or CaCl<sub>2</sub> as indicated. Chlorophyll concentration was 50  $\mu$ g/ml, temp. 25°C, pH 8.0. After a 5 min preillumination period (Stage A) the incorporation reaction was started by addition of ATP to a final concentration of 5 mM. After again 5 min in the light or the dark (Stage B) the reaction was terminated by addition of trichloroacetic acid (final concentration 4%), and <sup>32</sup> $P_i$  incorporation into the organic fraction determined essentially as in ref. 12. In Expt 1, less than 5% of the ATP was broken down by dark Mg<sup>2+</sup>—ATPase during the course of the exchange. Figures given in  $\mu$ moles  $P_i$  incorporated per h per mg chlorophyll.

| Stage     |                    | Divalent cation present |                  |      |
|-----------|--------------------|-------------------------|------------------|------|
| A         | В                  | Mg <sup>2+</sup>        | Ca <sup>2+</sup> | None |
| Experimen | t 1                |                         |                  |      |
| + DTE     | dark               | 6.2                     | 0                | 0    |
| + DTE     | light              | 10.3                    | 5.9              | 0.8  |
| Experimen | t 2                |                         |                  |      |
| - DTE     | light              | 10.4                    | 1.0              |      |
| + DTE     | light              | 18.8                    | 7.3              | _    |
| + DTE     | light + DCCD 10 µM | 15.1                    | 5.7              | _    |
| + DTE     | light + DCCD 50 µM | 5.7                     | 0.8              | -    |

the purified enzyme, but not in the latter. In that case, it should be possible to observe a Ca<sup>2+</sup>-stimulated turnover of the bound ATPase enzyme under conditions of an externally applied high-energy state.

In Table I it is shown that, indeed, under continuous illumination a Ca<sup>2+</sup>-stimulated phosphate incorporation into ATP can be induced in chloroplasts, with 40-60% of the activity of the Mg<sup>2+</sup>-stimulated ATP-P<sub>i</sub> exchange. Like the latter, it is greatly enhanced by preillumination in the presence of the di-sulphydryl compound dithioerythritol [1,9,10] and inhibited by dicyclohexyl carbodiimide (DCCD) (Expt 2); but whereas the Mg<sup>2+</sup>-stimulated exchange after induction continues in the dark by virtue of simultaneous ATPase activity [10], the Ca<sup>2+</sup>-stimulated reaction is strictly dependent upon continuous external energy supply in the form of light (Expt 1). As yet, it is not clear whether this Ca<sup>2+</sup>-stimulated phosphate incorporation represents a true ATP-P; exchange or an ATPase reaction followed by ATP-synthesis. In view of the absolute requirement for Mg<sup>2+</sup> in photophosphorylation, the latter possibility seems less probable, unless some firmly-bound Mg<sup>2+</sup> would suffice here. Besides, we were not able to show any significant Ca<sup>2+</sup>-ATPase activity in chloroplasts by varying the energy state of the chloroplasts over a wide range by background illumination and/or addition of uncoupler. The light-dependent Ca2+-ATPase found by Bennun and Avron [11] might however be relevant in this respect.

The important point is that under the proper energetic conditions also membrane-bound CF<sub>1</sub> exhibits Ca<sup>2\*</sup>-stimulated activity. The problem of the 'allotopy' of CF<sub>1</sub> towards different cations thereby is changed into a problem of the effect of divalent cations on energy conservation.

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